

Investigation of the electrospinnability of polymer solutions using tubeless siphoning

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Electrospinning

1. Electrospinning of polymer solution

- Electrospinning is a unique technique for fabrication of **nanofibers**
- Constant **elongational flow** of polymer solution is maintained from a reservoir
- An **extensional force** is applied by high electrical voltage
- A conductive substrate, normally grounded, serves as a **collector**

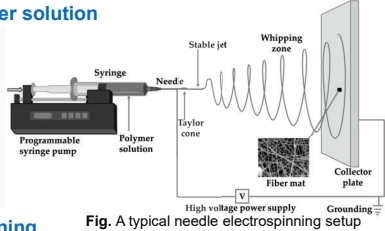
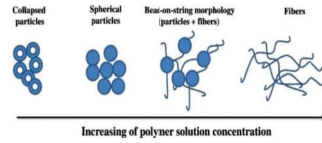


Fig. A typical needle electrospinning setup

2. Challenges of electrospinning

- Electrospinnable range specific to polymer
- Specific to molecular weight
- Polymer solution concentration
- Costly way of determining electrospinnable range



Increasing of polymer solution concentration

Tubeless siphoning

1. Tubeless siphoning of polymer solution

- Tubeless siphoning, a unique property of non-Newtonian fluid, is a **pure extensional flow**
- Unlike Newtonian fluid, siphoning action continues even when the siphon nozzle is raised above the free surface of the fluid
- Viscoelastic stresses, resulting from stretching of the polymer molecules in solution, support the weight of the column



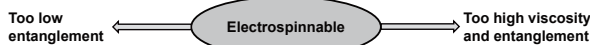
Fig. Tubeless siphoning of polyethylene oxide (PEO) solution

2. Factors influence Tubeless siphoning

- Molecular weight and concentration of polymer solution
- Entanglement of the polymer chains in the solution
- Applied extensional force

Motivation and objectives

- Electrospinning is possible for a concentration range of a specific polymer
 - Two low concentration → lack of chain entanglement → Does not form continuous fiber
 - Two high concentration → too high chain entanglement to draw the solution



Polymer concentration

- Tubeless siphoning for a polymer solution also possible for a definite range of concentration
 - Two low concentration → lack of chain entanglement → Does not form siphon column
 - Two high concentration → too high chain entanglement to form a siphon column
- Both the electrospinning process and tubeless siphoning involve **high extensional flow**

Objective of this work

- Establishing a correlation between siphonable range and electrospinnable range for a polymer solution with different molecular weight
- Predicting electrospinnability of a polymer solution from tubeless siphoning
- Reducing the costly process of determining electrospinnability

Materials and methods

1. Tubeless siphoning of PEO solutions

- A series of concentrations (0.05 -10 w/w %) of PEO solutions with MW 400k, 900k, and 5000k were investigated
- Backlight imaging technique was used for measuring the siphon height for each solutions

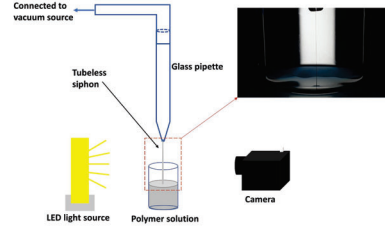


Fig. Tubeless siphoning setup for PEO solution with the arrangement of backlight imaging technique

Process parameters of electrospinning

Flow rate (ml/h)	Voltage (kV)	Needle tip-to-collector distance (cm)
0.5	15	14

2. Electrospinning of PEO solutions

- A needle electrospinning setup was used for electrospinning of all PEO solutions
- A constant flow rate, high electric voltage, and needle tip-to-collector distance were maintained

Results and discussion

1. Effect of concentration of PEO solution (400k) on siphon height

- Maximum siphon height was observed for 2% (w/w) solution
- Siphonable range : 0.25-6% (w/w)
- Pipette with diameter at or above 0.66 mm shows maximum siphon height
- Consistency of data for multiple runs (3-5) is remarkable

Force at pipette tip

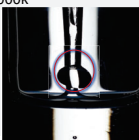
Pipette diameter (mm)	Cross sectional area in m ² (10 ⁻⁷)	Force in N (considering pressure -0.60 bar)
0.43	1.457	0.0087
0.58	2.628	0.0158
0.66	3.413	0.0205
0.87	5.959	0.0358
1.91	28.673	0.1720

2. Effect of molecular weight

- Siphon height increases significantly with increasing MW of PEO
- Siphon height for PEO 5000k is four times higher than that of PEO 400k
- Different maximum siphon heights concentration for different MW of PEO (400k: 2%, 900k: 2%, 5000k: 1.5%)
- Siphon heights for all runs (3-5) are remarkably consistent

3. Siphonable range of PEO solutions

- Jelly fish formation for PEO solutions with concentration at or above 2% for PEO 400k and 900k, and 1.5% for PEO 5000k
- Siphon height decreases significantly at jelly fish formation region
- No siphoning at concentration 6% for PEO 400k and 900k, and 5% for PEO 5000k



Jelly fish formation

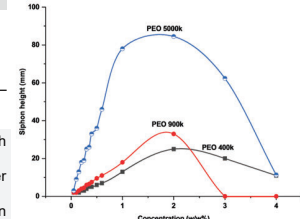
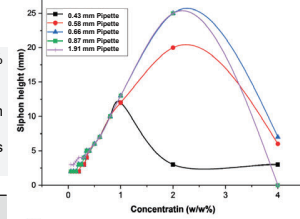


Fig. Suction pressure: ~ -0.60 bar, pipette diameter: 0.87 mm

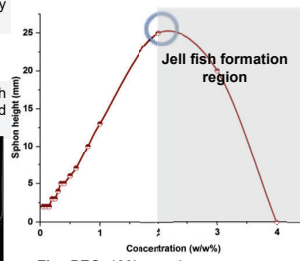


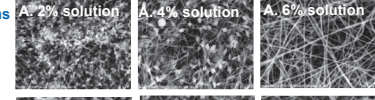
Fig. PEO 400k, suction pressure: ~ -0.60 bar, pipette diameter: 1.91 mm

Results and discussion

4. Electrospinnability of PEO solutions

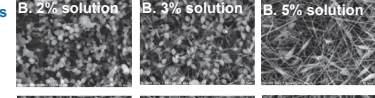
A. Electrospinnability of PEO solutions (MW: 400k)

- 2% solution: Beaded fibers
- 4% solution: Beaded fibers
- 6% solution: Uniform fibers



B. Electrospinnability of PEO solutions (MW: 900k)

- 2% solution: Beaded fibers
- 3% solution: Beaded fibers
- 5% solution: almost uniform fibers (fewer beads)



C. Electrospinnability of PEO solutions (MW: 5000k)

- 2% solution: Beaded fibers
- 3% solution: Beaded fibers
- 5% solution: Uniform fibers

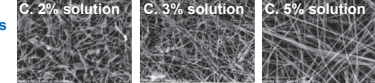
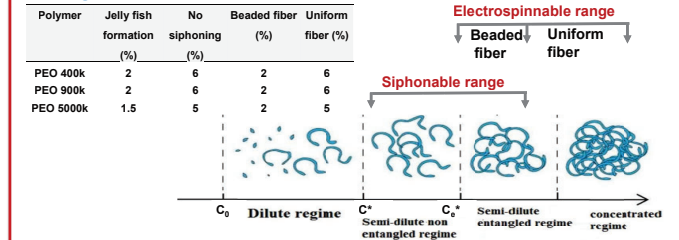


Fig. SEM images of electrospun fibers of PEO solution A: PEO 400k, B: PEO 900k, C: PEO 5000k

5. Correlation between siphonable range and electrospinnable range



Conclusions

- PEO solution exhibits tubeless siphoning within the range of 0.25 and 6%
- 'Jelly fish' structure along siphon column at 2% for PEO 400k and 900k, and 1.5% for PEO 5000k
- No siphoning was possible at 6% for PEO 400k and 900k, and 5% for PEO 5000k
- For PEO 400k and 900k: beaded fiber at concentration 2-5%
bead free fibers at or above 6%
- For PEO 5000k: beaded fiber at concentration 1.5-4%
bead free fibers at 5%

Future work

- Correlation between siphonable range and electrospinnable range for other water-soluble polymers such as polyacrylamide (PAM) and PAM-co-DADMAC
- Effect of addition of salts, surfactants, and PEG, temperature, pressure and solvents on tubeless siphoning
- Measuring extensional viscosity of polymer solutions from maximum siphon height measurement
- Using backlight imaging technique to correlate siphon pattern with solution concentration and pipette diameter

References

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